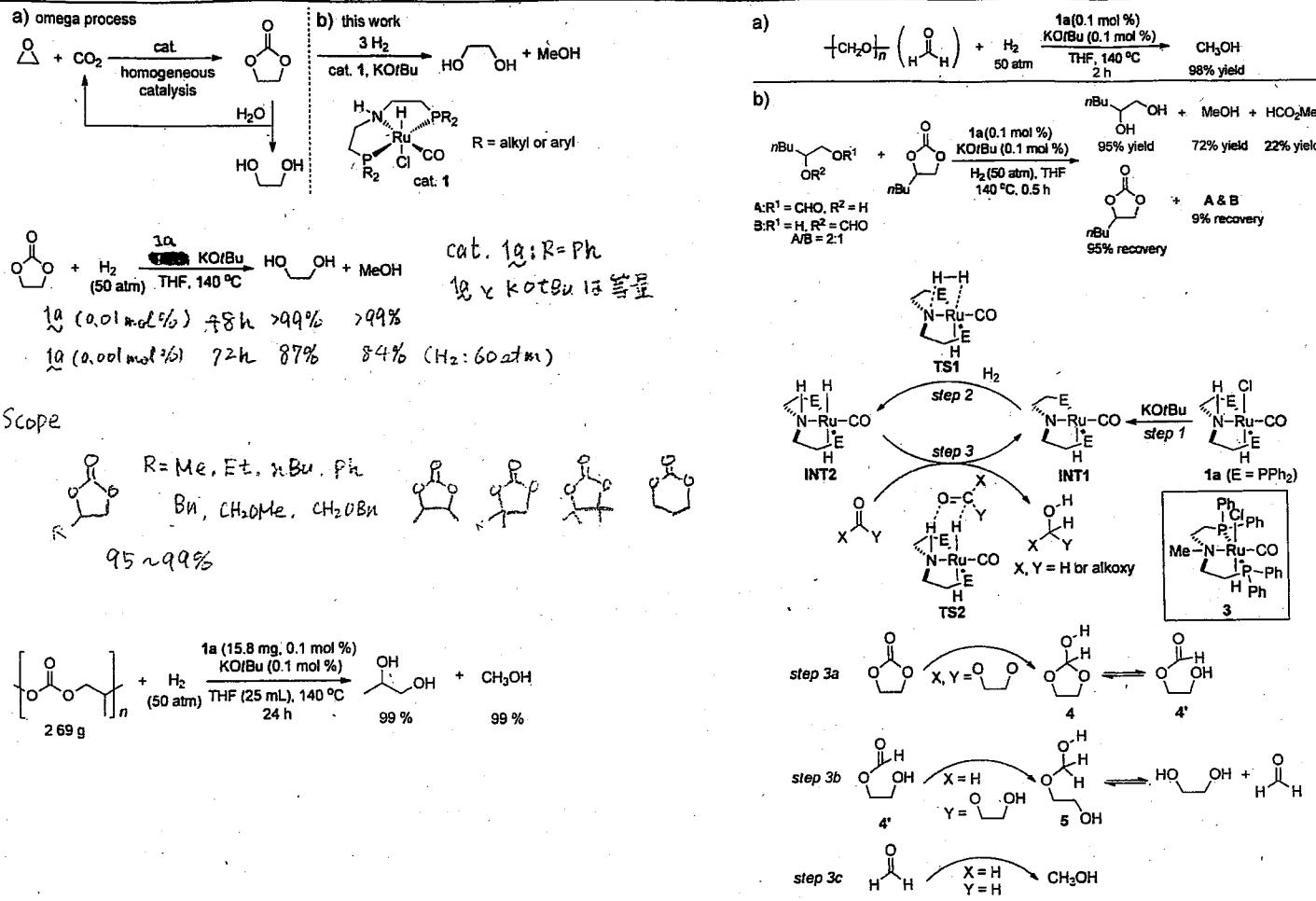
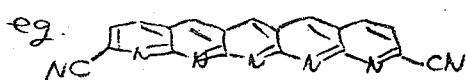


Catalytic Hydrogenation of Cyclic Carbonates: A Practical Approach from CO₂ and Epoxides to Methanol and Diols

Qichun Zhang	Nanyang Technological University, Singapore	J. Am. Chem. Soc. ASAP (doi: 10.1021/ja310131k1)	Youhei Takeda
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Synthesis and Physical Properties of Four Hexazapentacene Derivatives

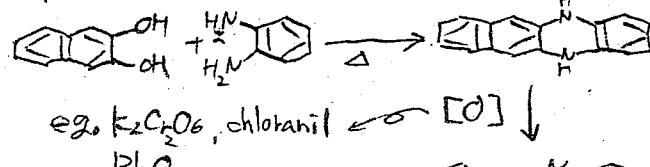
• aza acenes: candidates for n-type organic materials



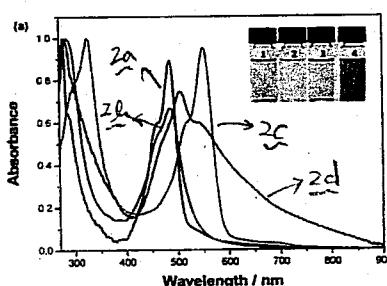
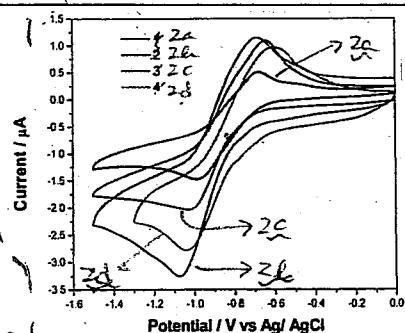
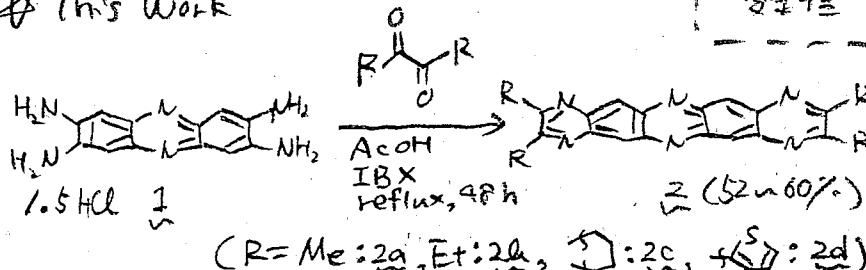
EA 3.21 eV 3 < EA < 4 eV (理想)

JACS 2007, 129, 1805.

• 徒手合成法 (Pure Appl. Chem. 2010, 953.)



• This Work



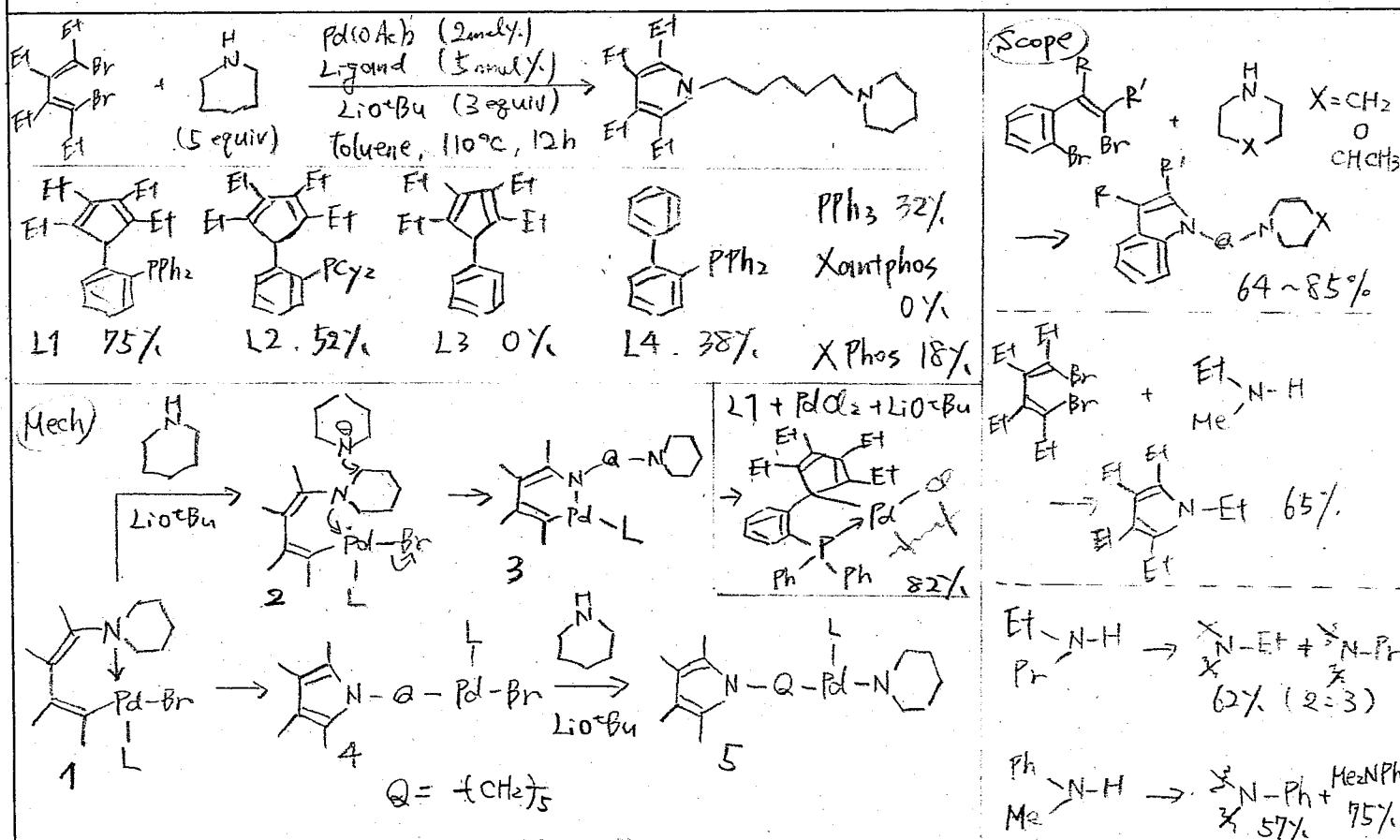
UV absorption spectra
in DMF

Compd	HOMO/eV	LUMO/eV	ΔE/eV
2a	-6.12	-3.57	2.55
2b	-6.08	-3.53	2.55
2c	-6.03	-3.58	2.45
2d	-5.84	-3.59	2.25

cf.

pentacene	-5.14	-3.37	1.77
hexacene	-4.96	-3.56	1.57

Cyclopentadiene-Phosphine/Palladium-Catalyzed Cleavage of C–N Bonds in Secondary Amines:
Synthesis of Pyrrole and Indole Derivatives from Secondary Amines and Alkenyl or Aryl Dibromides



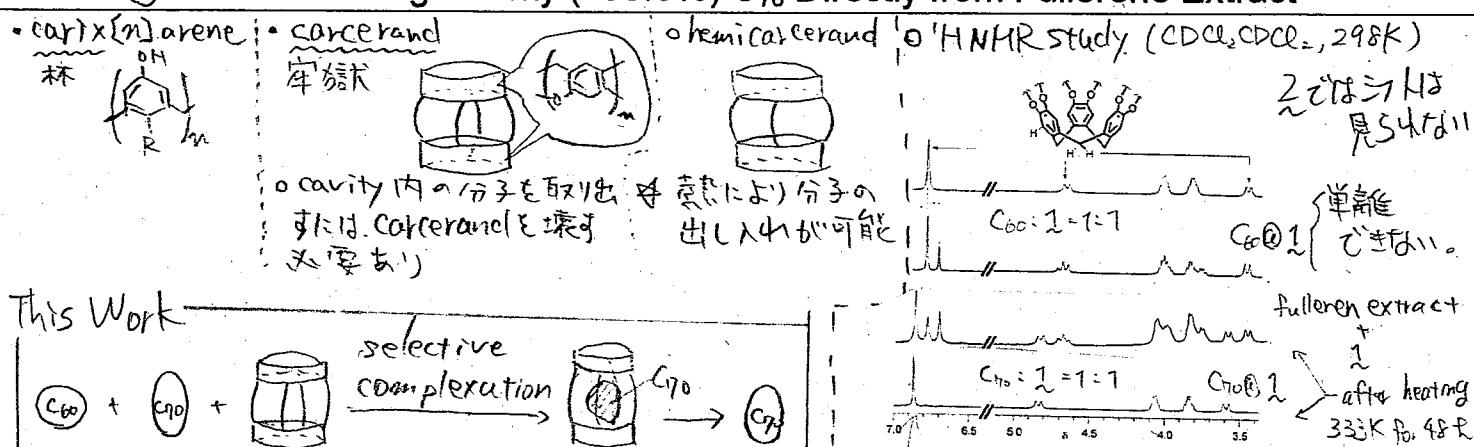
Chiu Sheng-Hsien

National Taiwan University

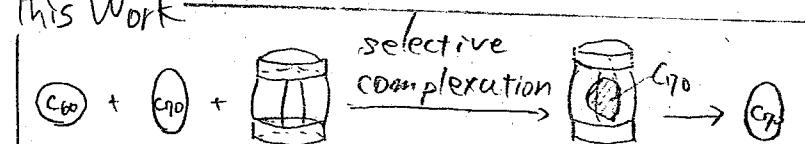
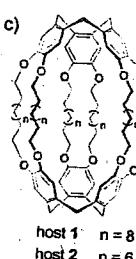
013027957

Nagamachi

Hemicarcerand Formation With a Cyclotrimeratrylene-Based Molecular Cage Allows Isolation of High-Purity ($\geq 99.0\%$) C_{70} Directly from Fullerene Extract



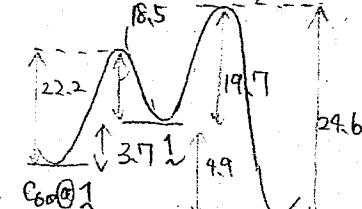
This Work

 C_{70} @ Host Molecule

7 steps.

total yield:
 $m=8$: 0.8%
 $m=6$: 1.4%

Constrictive and intrinsic bond energy (kcal/mol)



association rate constant $k = 1.026 \text{ M}^{-1}\text{s}^{-1}$

$\text{C}_{60}\text{@}1$: $K_a = 4200 \text{ M}^{-1}$

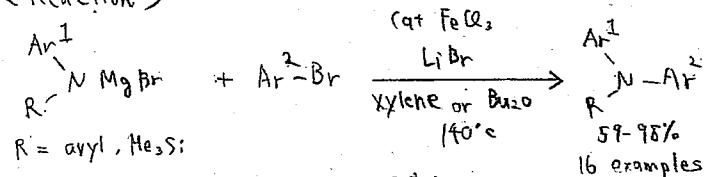
$\text{C}_{70}\text{@}1$: $K_a = 500 \mu\text{M}^{-1}$

Selective separation of C_{70} from fullerene extract
 C_{70} + fullerene extract (300mg) \downarrow in toluene at 303K, 12h
 $\text{C}_{60}:\text{C}_{70}:\text{others} = 68:25:7$
centrifuge \downarrow in CHCl_3
 \downarrow CHCl_3 at 313K, 16h
centrifuge \downarrow 濃縮
 \downarrow 濃縮
 \downarrow 濃縮
 \downarrow 濃縮
pure C_{70} (6.5mg) (99.0%)

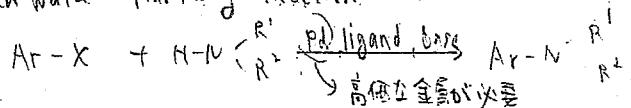
○不純物は $\text{C}_{76}\text{@}1$ が主で
オストラムを変えて higher fullerene など
を選択的に抽出できる可能性あり!!

Iron-Catalyzed Aromatic Amination for Nonsymmetrical Triarylamine Synthesis

<Reaction>

<Previous work> (MPPNP₃-9, 合成)

Buchwald - Hartwig Reaction.



<Optimization>

entry	catalyst (X mol%)		time	yield (%)	irrecoverable
	(Y equiv)	additive (Y equiv)			
1	FeCl ₃ (5)	none	24	10	89
2	FeCl ₃ (5)	LiBr (0.2)	24	26	57
3	FeCl ₃ (5)	LiBr (2.0)	12	51	41
4	FeCl ₃ (5)	LiBr (4.0)	72	>99	0
5	FeCl ₃ (0.5)	LiBr (4.0)	48	99	0

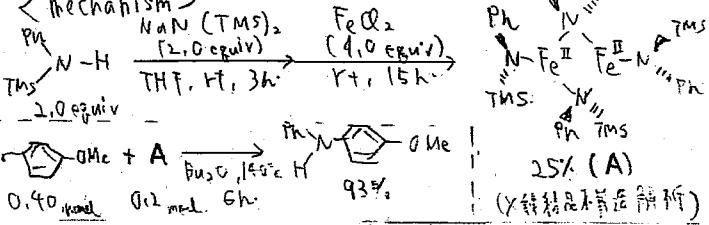
* FeCl₃ R² ffpm たゞ、
CuCl₂, FeCl₃ (0.5 mol%) 用いても反応は進行しない。 Ni(O₂)₃ (0.5 mol%)
ZnCl₂ 88% の收率で反応が進行した。 FeCl₃ 100% 收率で反応が進行した。
N. の量 0.0035 mol% まで触媒量を減らすと收率が 87% まで低下した。

<Scope>

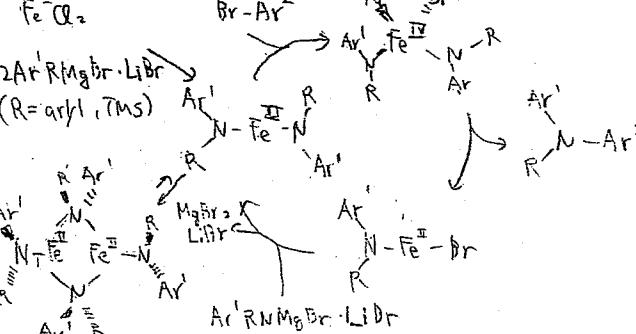
 $\text{Ar}^2 \cdots \text{Ar}^1 \text{N}(\text{Ar}^2) \text{Ar}^1$

アリル、アトニル、ニトリル部位が存在すれば、目的反応は円滑に進行せず、複雑な混合物を生ずる。

<Mechanism>



(Yield of product A)



Ito Hajime

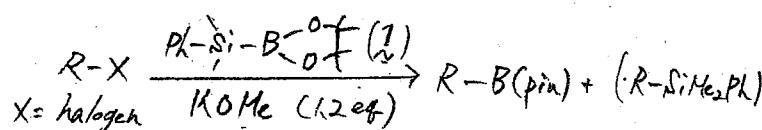
Hokkaido University

JACS DOI: 10.1021/ja309578k

Okumura

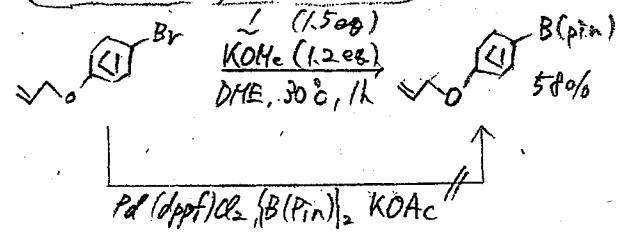
Anomalous Reactivity of Silylborane: Transition-Metal-Free Boryl Substitution of Aryl, Alkenyl, and Alkyl Halides with silylborane/Alkoxy Base Systems

This Work

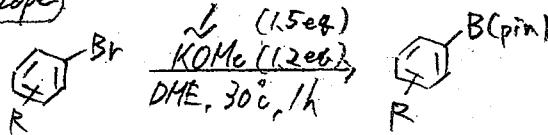


R = alkyl, alkenyl, aryl.

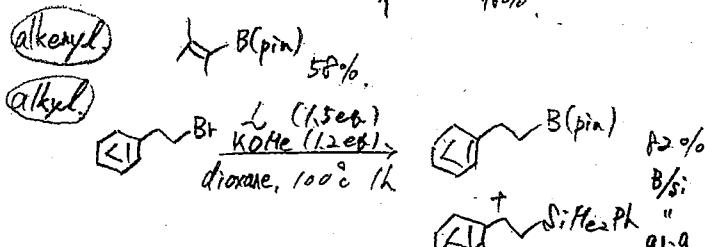
(Chemoselective Boryl Substitution)



Scope

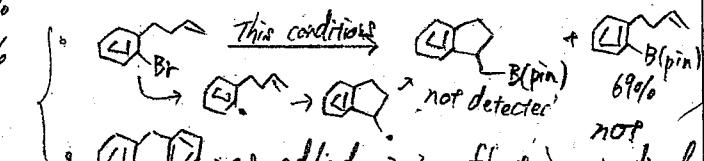


R = X-OMe, 77%, B/Si = 95:5
 (Ar) = H, 68%, X-F, 63%, X-Cl, 56%
 = 2-Me, 67%, 3-Me, 62%, X-Me, 68%
 = X-CF₃, 50%.

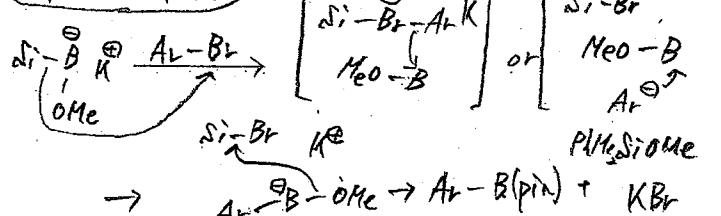


Mechanistic Study

• hew. reagents \rightarrow no effect \rightarrow Metal-Free-Reaction



Proposed mechanism



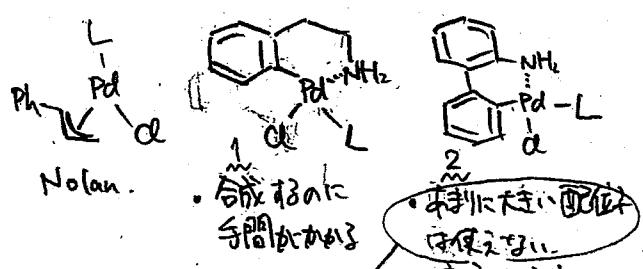
Design and preparation of new palladium precatalysts for C-C and C-N cross-coupling reactions

Pd sources

$\text{Pd}_2(\text{dba})_3$: Pd nanoparticles & free dba 含有。
dba が 反応に 影響 し 得る。

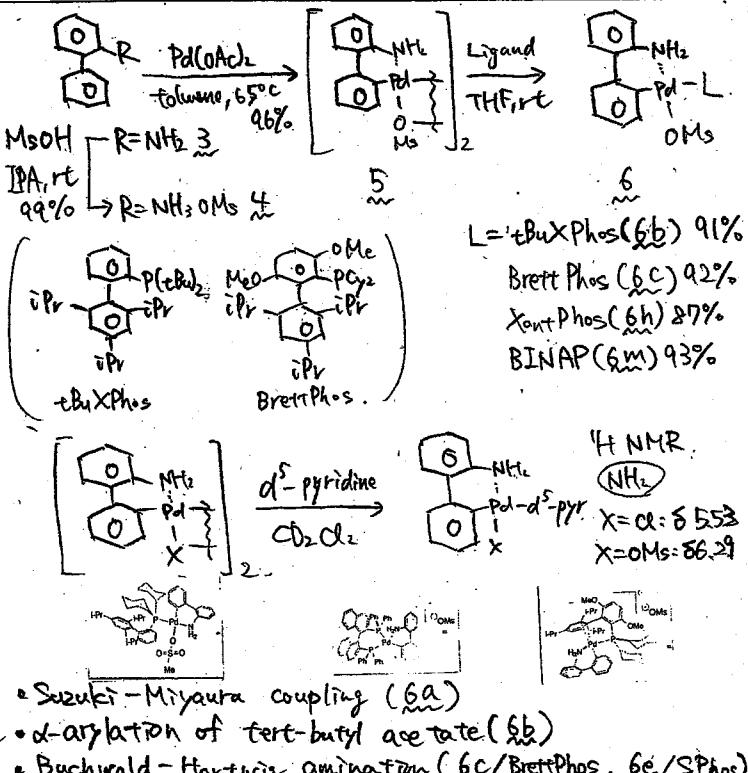
$\text{Pd}(\text{OAc})_2$, PdCl_2 : $\text{Pd}^{(0)}$ は 還元 可能 で ある。

→ 解決法: Pre-ligated Pd(II) sources
(air and moisture stable)



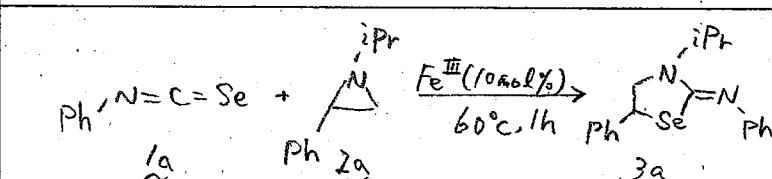
in a Cl-E (電子求引性の利点) が ある。

OMs ($\text{O}-\text{S}-\text{Me}$) が 最適 である。



- Suzuki - Miyaura coupling (6a)
 - α -arylation of tert-butyl acetate (6b)
 - Buchwald - Hartwig amination (6c/BrettPhos, 6e/SPhos)
- * 2-iodotetrafluoroethane \rightarrow 2^a a Pd(II) source \rightarrow 2^b a Pd(II) source \rightarrow 2^c a Pd(II) source \rightarrow 2^d a Pd(II) source

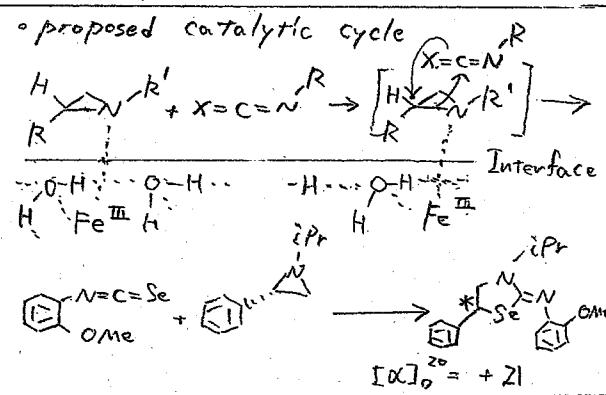
“On Water”: Efficient Iron-Catalyzed Cycloaddition of Aziridines with Heterocumulenes



Optimization

	Catalyst (Fe ^{III})	Solvent = H ₂ O
$\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$	100%	
$\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$	87%	
$\text{Fe}_2(\text{SO}_4)_3 \cdot 5\text{H}_2\text{O}$	58%	
$\text{Fe}(\text{acac})_3$	47%	
no catalyst	12%	

Aziridine + heterocumulenes
catalyst or stoichiometric reagent
 $\text{Pd}, \text{Ni}, \text{HBF}_4, \text{Mg-MeOH}, \text{PBu}_3, \text{NaI}$



Scope

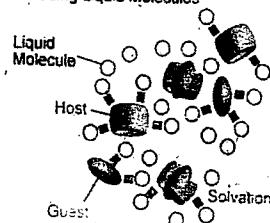
isose/enoicyanate ($\text{R}'-\text{N}=\text{C}=\text{Se}$)	aziridine ($\text{R}'-\text{N}^{\text{a}}$)	heterocumulene ($\text{R}'-\text{N}=\text{C}=\text{X}$)
$\text{R}' = -\text{C}_6\text{H}_4\text{X}$ ($\text{X} = -\text{OMe}$ (scale ok), -I, $-\text{NO}_2$ etc.)	$\text{R}' = \text{H}, \text{Alkyl}, \text{Ph}$ Bz, Boc (not n.d.)	$\text{Ar}'-\text{N}=\text{C}=\text{n}$ $\text{Ph}'-\text{N}=\text{C}=\text{O}$ $\text{Ph}'-\text{N}=\text{C}=\text{S}$
1-naphthyl 2-fluorophenyl cyclohexyl	$\text{R}' = \text{Ar}$	$\text{Ar}'-\text{N}=\text{C}=\text{n}$ $\text{Ar}'-\text{N}=\text{C}=\text{O}$ $\text{Ar}'-\text{N}=\text{C}=\text{S}$
61 ~ 93%	61% ~ 91%	65% ~ 79%

Cyclic Host Liquids for Facile and High-Yield Synthesis of [2]Rotaxanes

Liquid Host Molecules (This Work)

Strategy

a. Using Liquid Molecules

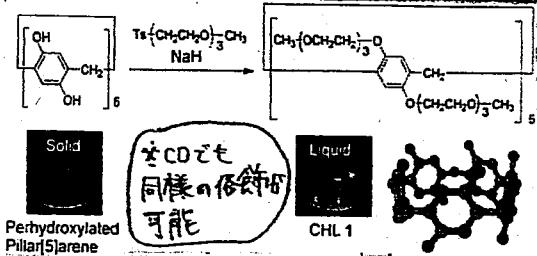


b. Using CHL



従来の H-G system → 溶媒和により、H-G の接合が妨げられる。

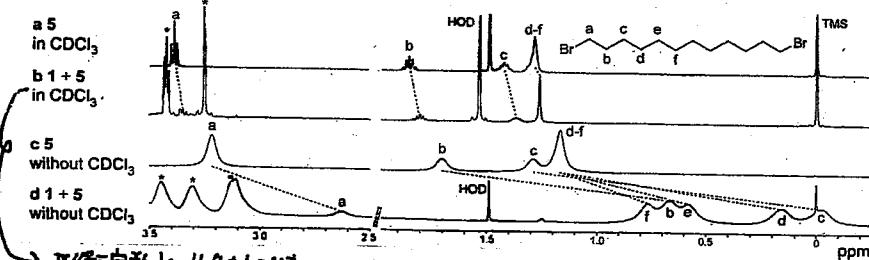
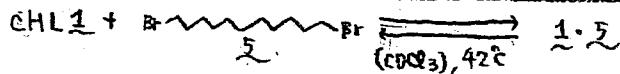
Synthesis of CHLs (=Cyclic Host Liquids)



TGA & DSC: -50°C ~ 250°C で 液相を保つ
TEO 鎮か両親媒性をもつた。様々な溶媒と混和する!

(O water, alcohols, acetone, CHCl3, ethers, toluene, DMF, DMSO × Hexane)

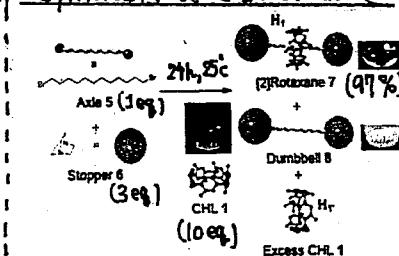
Complexation between CHL 1 and 1,12-dibromododecane 5



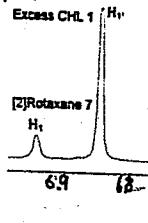
$$\text{平衡定数 } K = 16.9 \pm 1.0 \text{ M}^{-1}$$

Neat 条件(c,d)である場合の方が複合化が効率的に進行!!

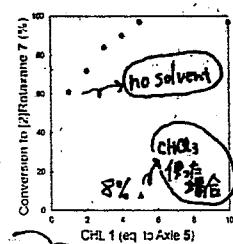
Synthesis of [2]Rotaxane



¹H NMR



plot of Conversion to 7



8: acetone 洗浄で除去
9: Et2O, Hexane を用いた洗浄により
CHL を用いれないと効率的に
[2]Rotaxane が合成可! AC!

John F. Hartwig et al.

University of California, Berkeley

JACS

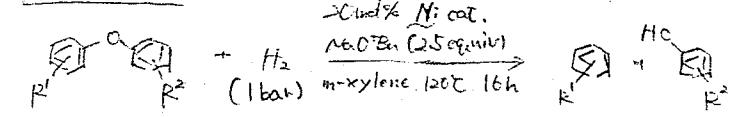
DOI: 10.1021/ja02081912

M1 稼働

A Heterogeneous Nickel Catalyst for the Hydrogenolysis of Aryl Ethers without Arene Hydrogenations

Differences of Ni catalysts

Conditions



~20 mol % Ni cat. (work)
substrate

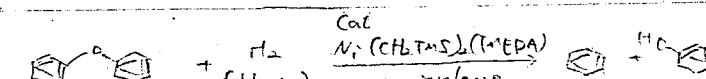
62% (48h: 99%)

100%

41%

97%

(100% conversion, A) + 10



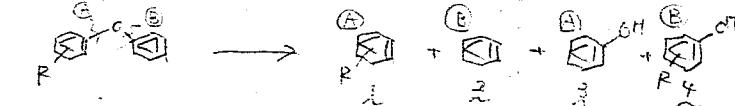
20 mol % cat., 120°C, 16 h → > 95% yield

2 mol % cat., 120°C, 96 h → > 95% yield

2 mol % cat., 140°C, 24 h → 95% yield

Scope ($\text{Ni}(\text{C}_6\text{H}_5\text{SiMe}_3)_2(\text{TMEDA})\text{cat}$)

unsymmetrical diaryl ethers



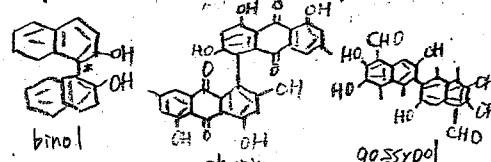
R	Ni (mol %)	time (h)	Z (%)	Z (%)	Y (%)
o-	10	96	22	60	17
m- Me	5	48	18	82	17
p- R	5	48	13	85	11
p- ^t Bu	5	48	3	90	90
c- OMe	0.5	48	45	54	44
m- OMe	10	48	21	46	22
p- OMe	10	48	15	2	18
c- OH	5	48	80	9	80
m- OH	10	48	75	8	75
p- OH	15	41	97	-	97

benzyl ethers

Substrate	Ni (mol %)	ArCH ₂ / %	Substrate	Ni (mol %)	ArCH ₂ / %
<chem>*c1ccc(cc1)Oc2ccccc2</chem>	2	98	<chem>*c1ccc(cc1)Oc2ccccc2</chem>	0.25	93
<chem>*c1ccc(cc1)Oc2ccccc2</chem>	2	87	<chem>*c1ccc(cc1)Oc2ccccc2</chem>	20	99

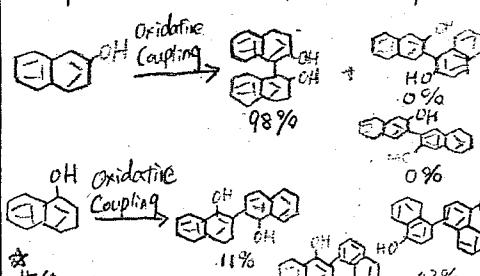
The Acetal Concept: Regioselective Access to ortho, ortho'-Diphenols via Dibenzo-1,3-dioxepines

< Representative ortho, ortho'-diphenols >

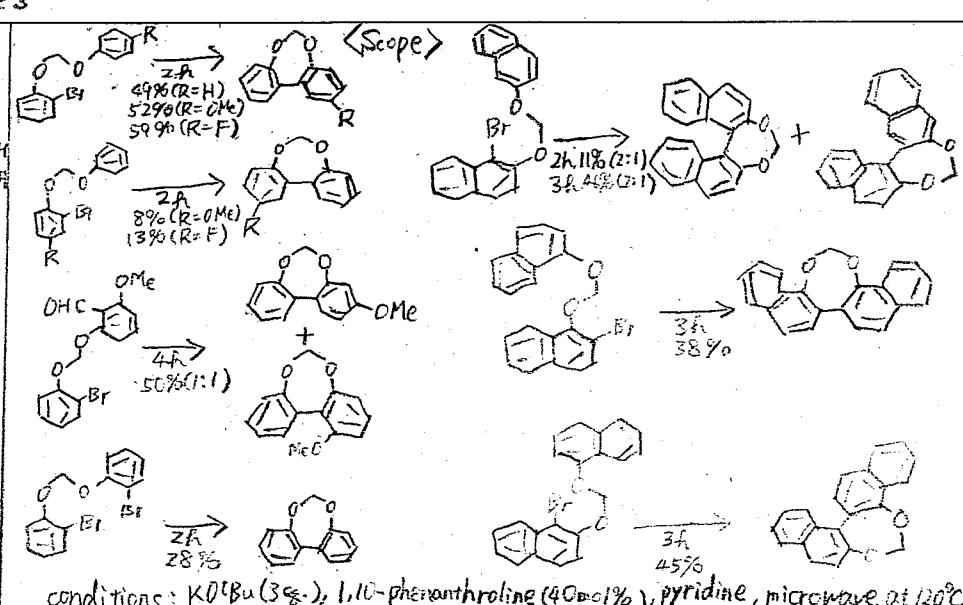
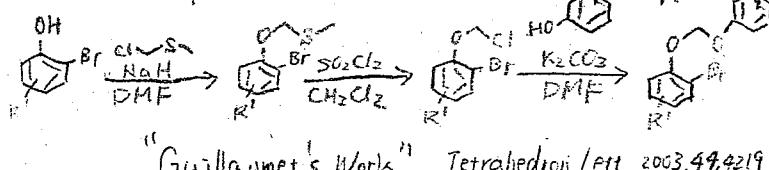


○ 天然物、薬、不育配位子など数多く存在する。

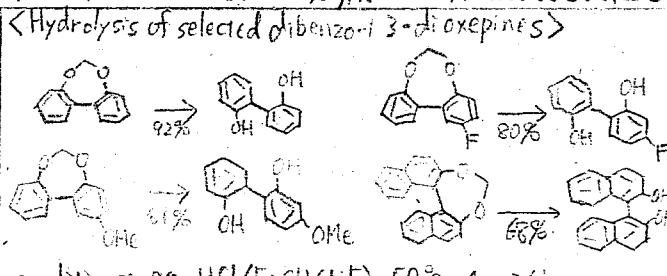
< Comparison of the dimerization of naphthols >



< Acetal concept >



conditions: $\text{KOTBu}(3\text{ eq.}), 1,10\text{-phenanthroline}(40\text{ mol\%}), \text{pyridine}, \text{microwave at } 120^\circ\text{C}$



Takeaki Iwamoto

Tohoku University (Japan)

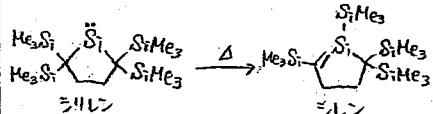
JACS
(10.1021/ja310391m)

M1 西田

New Isolable Dialkylsilylene and its Isolable Dimer That Equilibrate in Solution

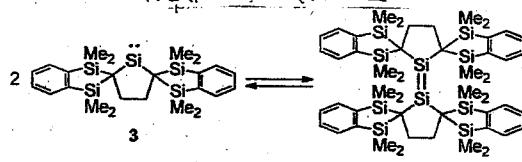
Silylenes (R_2Si)

…カルベンへ類縁体。
様々な反応性が期待できる。
しかし、安定性にえぐく合成が難しう。
※無置換シリレンの単離は過去に一例のみ



Ishida, S. et al., Organometallics, 2009, 28, 919.

今回、新たに無置換シリレンの単離を達成

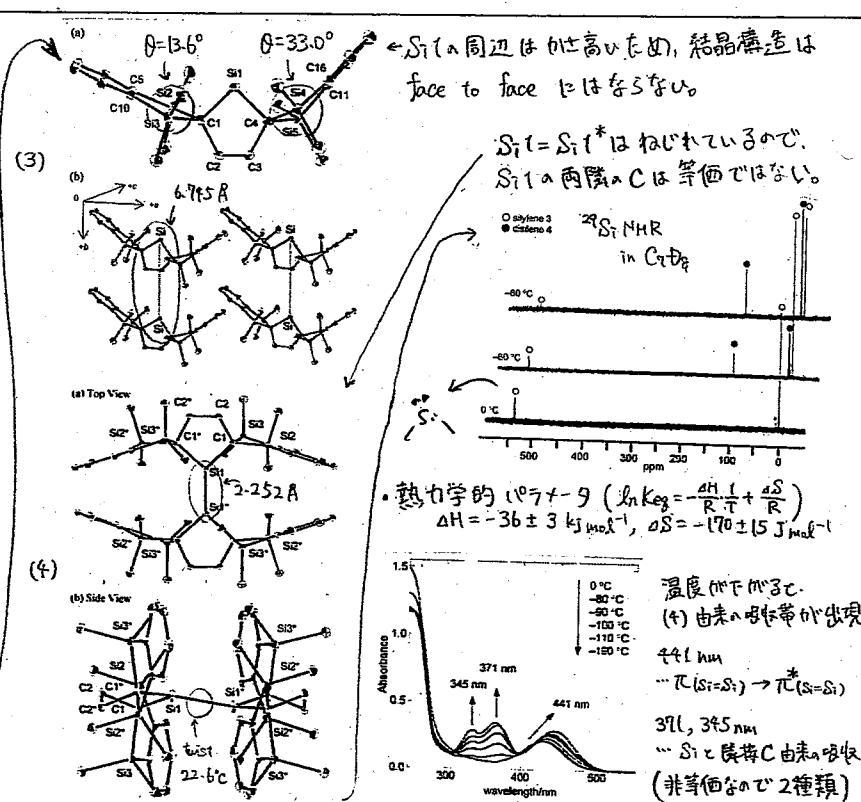


溶液状態で dimer (4) と平衡

monomer (3) を再結晶 ($\text{HgF}_2, -30^\circ\text{C}$)

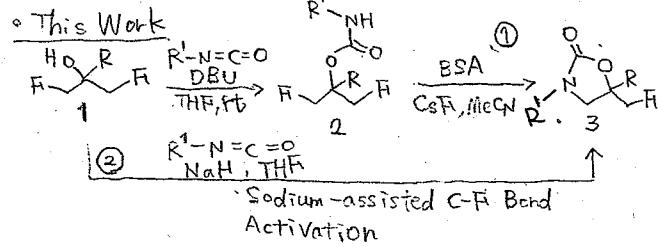
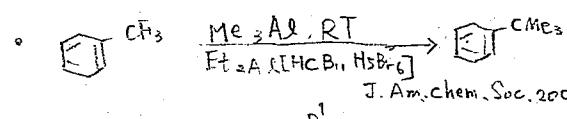
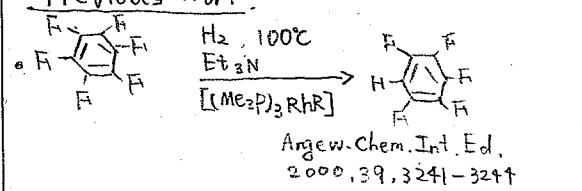
↓

{ monomer (3): yellow crystal
dimer (4): orange-red crystal (微量) }

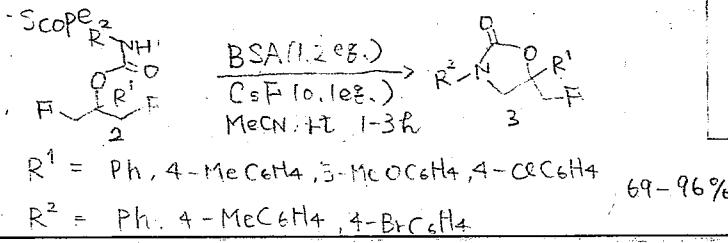


C-F Bond Activation of Unactivated Aliphatic Fluorides: Synthesis of Fluoromethyl-3,5-diaryl-2-oxazolidinones by Desymmetrization of 2-Aryl-1,3-difluorodiboran-2-ols.

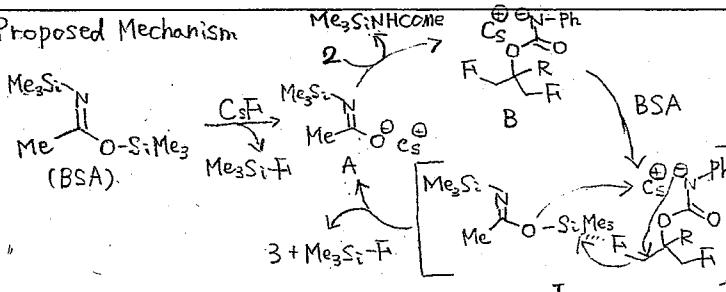
Previous Work:



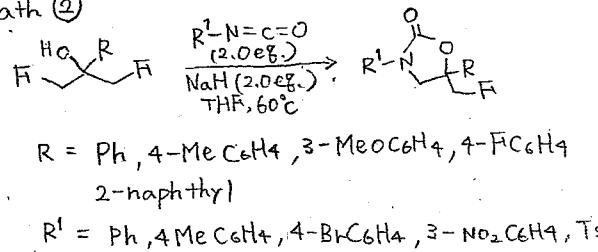
Path ① (2 → 3)



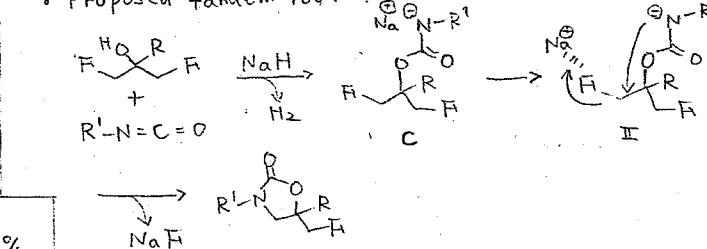
Proposed Mechanism:



Path ②



Proposed tandem route:



Yao Fu

University of
Science and Technology of ChinaAngew. Chem. Int. Ed
DOI: 10.1002/anie.201206681

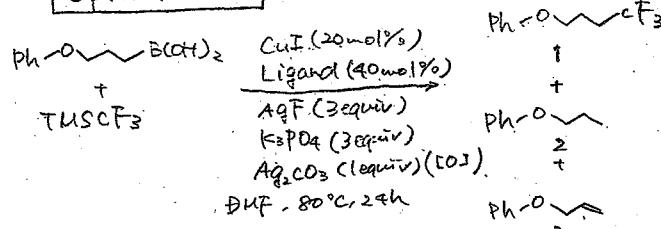
B4 黑田

Copper-Promoted Trifluoromethylation of Primary and Secondary Alkylboronic Acids

Trifluoromethylation

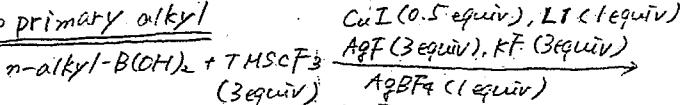
- Transition-metal-catalyzed (Pd, Cu)
- sp^2 -, sp - carbon
- allylic sp^3 - carbon (recent studies)

Optimization:



This Work:

- nonactivated sp^3 - hybridized carbon
- primary alkyl

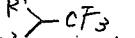


- secondary alkyl
- $R_2\text{C}\text{-B(OH)}_2 + \text{TMSCF}_3$ (3 equiv) $\xrightarrow{\text{L1 (40 mol\%)}, \text{AgF (3 equiv)}, \text{KF (3 equiv)}, \text{AgOTs (1 equiv)}}$ $\xrightarrow{\text{DMF}, 50^\circ\text{C}, 24\text{ h}}$



36~78%

14 examples



39~54%

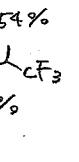
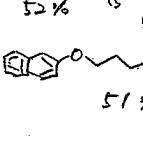
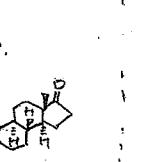
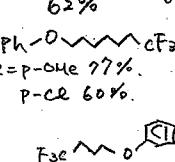
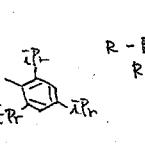
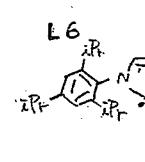
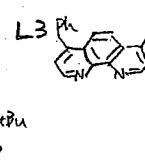
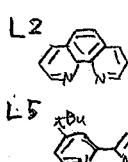
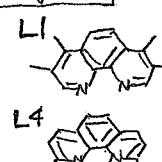
7 examples

L1. 26/24/11% (1/1/3)

L2. 16/25/8% L3 18/25/9%
L4. trace/18/5% L5 14/16/10%
L6. trace/16/4% (GC yield)

$\rightarrow \text{L1, [L1]} = \text{AgBF}_4 \text{ (1~2 mol\%)}$
 $\text{[L1]}: \text{Ag}_2\text{O}, \text{AgOAc}, \text{AgOTf}, \text{AgBF}_4, \text{BQ}$
trace trace 16% 50% trace
 $\rightarrow \text{L1, [L1]} = \text{AgBF}_4 \text{ (1~2 mol\%)}$
base = $\text{K}_2\text{CO}_3, \text{Cs}_2\text{CO}_3, \text{BuOK}, \text{KF}$
38% 55% 82% 68% → 88%

Ligand:



Scope:

Primary

Secondary

Secondary

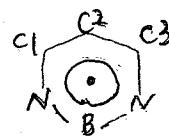
Secondary

Synthesis and Characterization of β -Heterocyclic π -Radical

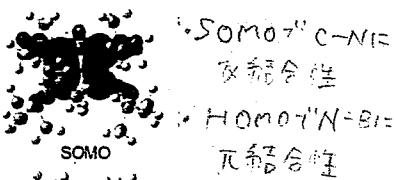
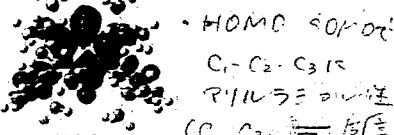
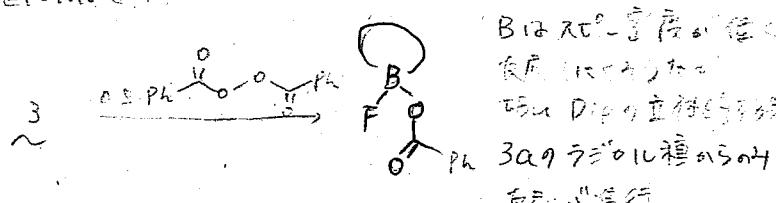
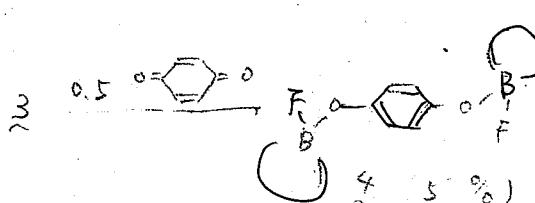
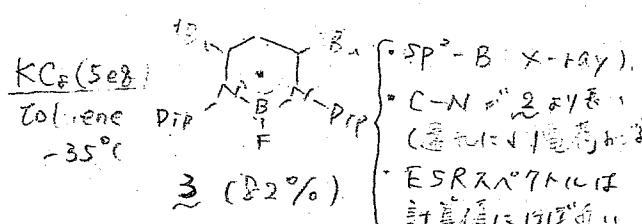
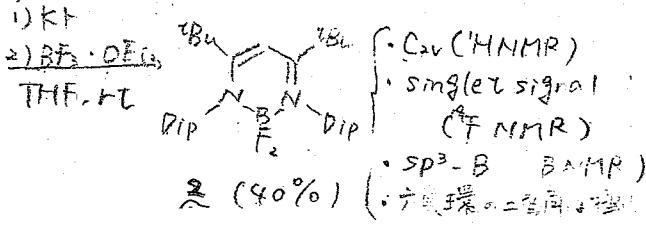
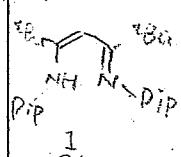
and Its Reactivity as a Hückel Radical.

・ホウ素中心のラジカルは5電子不安定

・リス塩基などによる安定化法

ex) NHC-BR_2 , $\text{RC}_2\text{HgN-BH}_2$ 

<DFT計算>

• SOMO $^{\pm}$ C-N \equiv
反結合性• HOMO $^{\pm}$ N-B \equiv
π結合性• HOMO $^{\pm}$ C \equiv O $^{\pm}$
C₁-C₂-C₃π
アーチラニン性(C₁-C₃)π
π共役• 2つのLUMO $^{\pm}$ B中にと大きな軌道相互作用

BはRC $^{\pm}$ -官能基の活性化に与効果をもつ
Dipは立従性をもつ
Ph 3aがラジカル種の3a2で反応が進行